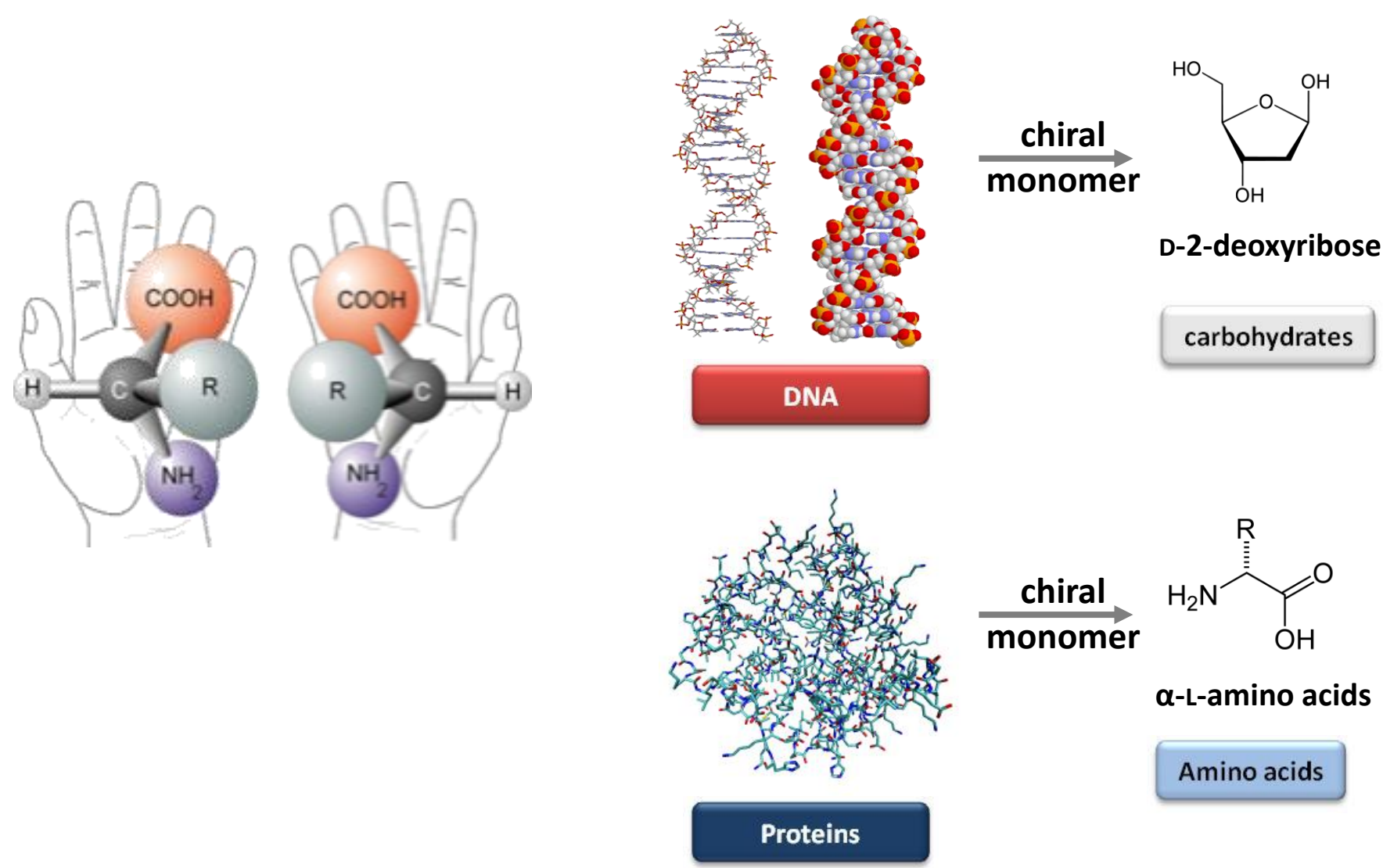


A conceptual illustration of the origin of life. A human hand holds the Earth, which is being struck by a comet. From the impact site, a trail of molecular models extends across the sky. At the bottom, a spectral line graph is shown with several peaks, some of which are highlighted with molecular models above them. The background is a dark, starry space.

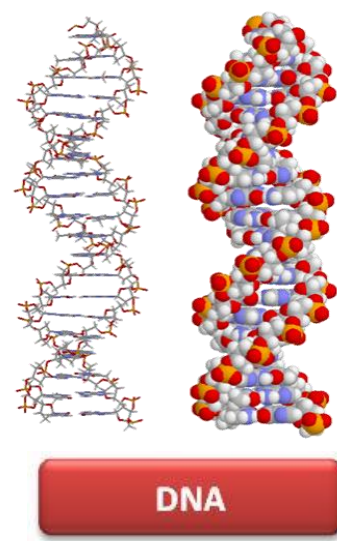
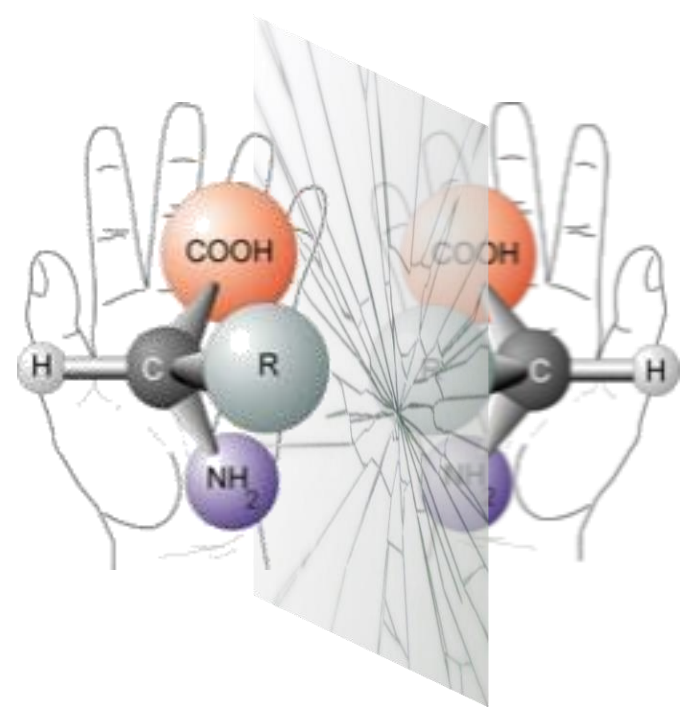
ORIGIN OF LIFE
HOW LIFE BECAME HANDED?

Cornelia MEINERT, *Institut de Chimie de Nice*, Université Côte d'Azur
Rencontres SFE 2018 à l'Observatoire de Haut Provence, October 16th, 2018

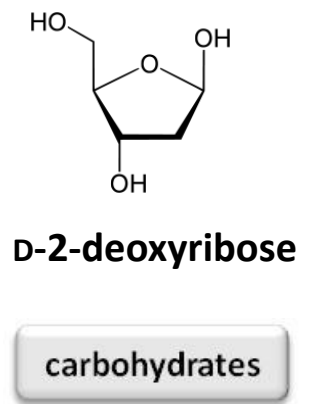
HOMOCHIRALITY OF LIFE



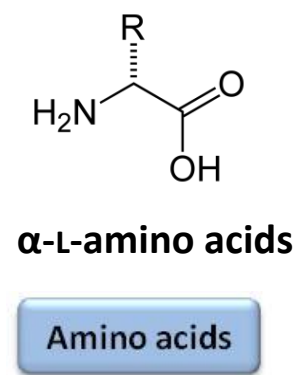
HOMOCHIRALITY OF LIFE



chiral
monomer

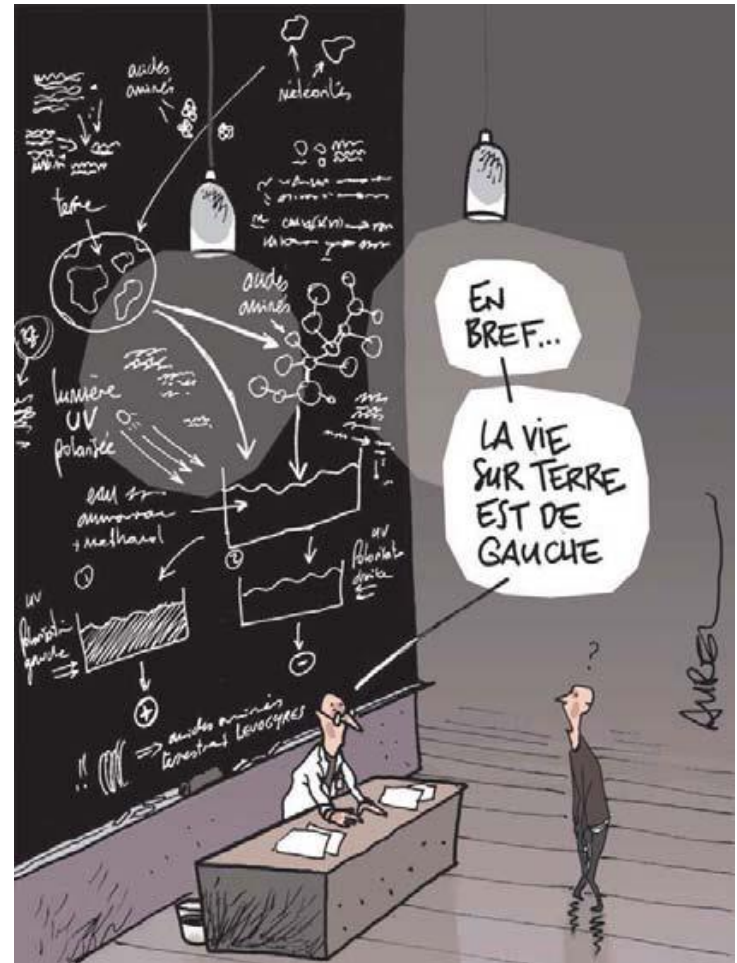


chiral
monomer



How did life originate ?

And why were left-handed amino acids
& right-handed sugars selected for its
architecture ?



« La lumière a donné un sens à la vie »
Le Monde: 08-01-2011

Prebiotic chemicals—amino acid and phosphorus—in the coma of comet 67P/Churyumov-Gerasimenko

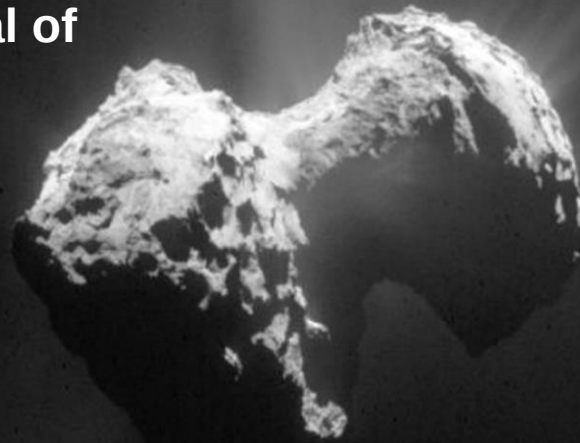
Kathrin Altwegg,^{1,2*} Hans Balsiger,¹ Akiva Bar-Nun,³ Jean-Jacques Berthelier,⁴ Andre Bieler,^{1,5} Peter Bochsler,¹ Christelle Briois,⁶ Ursina Calmonte,¹ Michael R. Combi,⁵ Hervé Cottin,⁷ Johan De Keyser,⁸ Frederik Dhooghe,⁸ Bjorn Fiethe,⁹ Stephen A. Fuselier,¹⁰ Sébastien Gasc,¹ Tamas I. Gombosi,⁵ Kenneth C. Hansen,⁵ Myrtha Haessig,^{1,10} Annette Jäkel,¹ Ernest Kopp,¹ Axel Korth,¹¹ Lena Le Roy,² Urs Mall,¹¹ Bernard Marty,¹² Olivier Mouis,¹³ Tobias Owen,¹⁴ Henri Rème,^{15,16} Martin Rubin,¹ Thierry Sémon,¹ Chia-Yu Tzou,¹ James Hunter Waite,¹⁰ Peter Wurz¹

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10.1126/sciadv.1600285

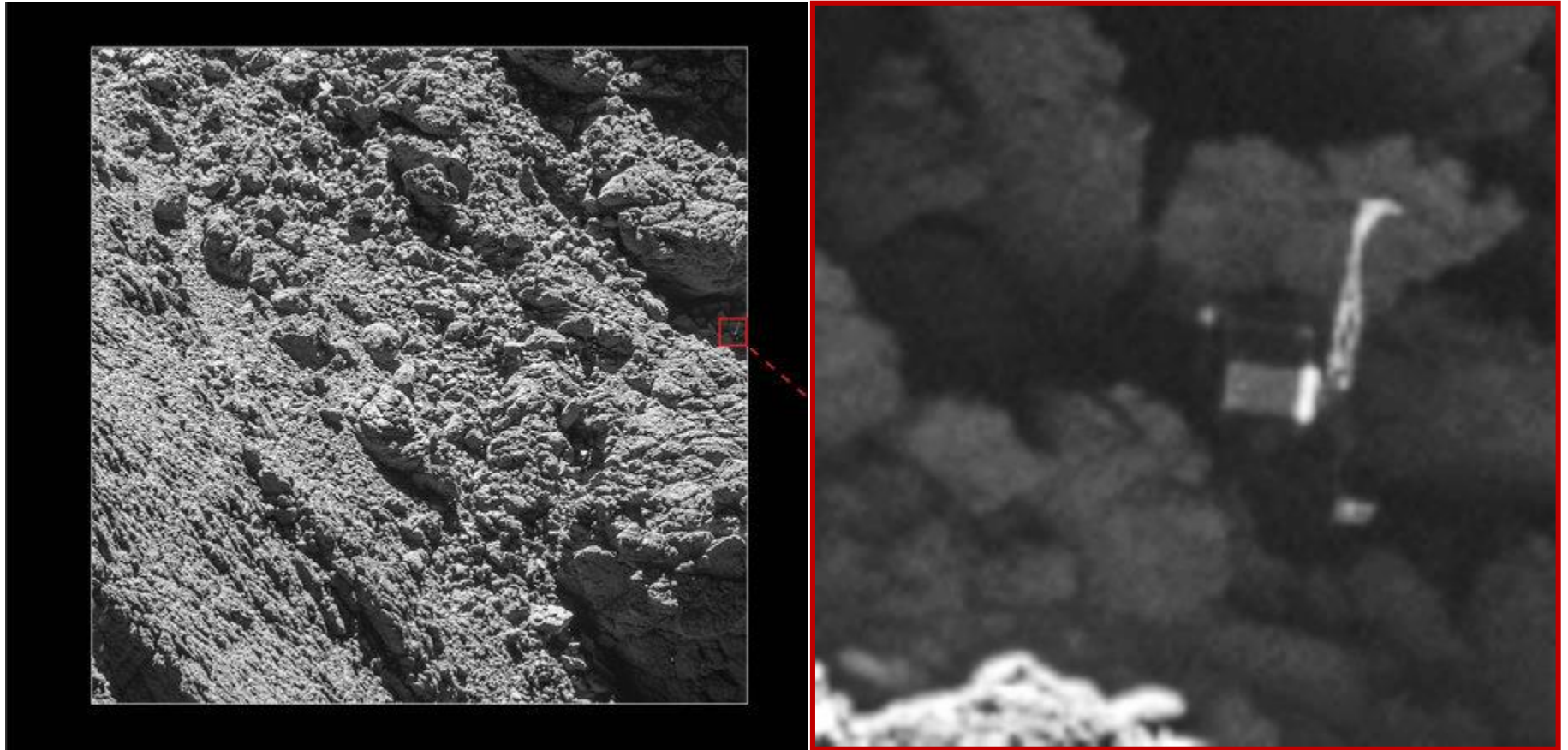


**Phosphorus and glycine recently
detected in the 67P comet !**

**These prebiotic ingredients are
present in the pristine material of
our solar system**



PHILAE FOUND!



The images were taken **on 2 September 2016** by the OSIRIS narrow-angle camera as the orbiter came within 2.7 km of the surface.

BUILDING BLOCKS OF LIFE: WHERE DO THEY COME FROM?



Murchison meteorite contains over 500 organic molecules:

Hydrocarbons	> 20 ppm
Aldehydes and ketones	> 10 ppm
Alcohols	> 10 ppm
Amines	> 1 ppm
Carboxylic acids	> 100 ppm
Sulfonic acids	> 100 ppm
Phosphonic acids	> 1 ppm
Purines & Pyrimidines	> 1 ppm
Amino acids	> 10 ppm

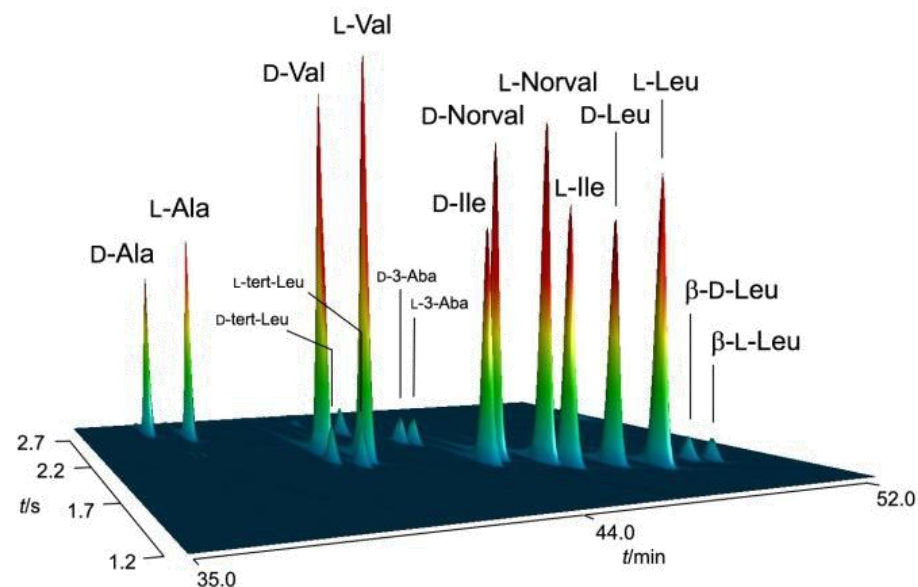
Cronin J.R.: In: Brack, A., The Molecular Origins of Life.
Cambridge University Press 1998, pp. 119-146.



Cosmic Evolution of elements and organic molecules.

Table 1 | Enantiomeric excesses ee_L in the Murchison meteorite.

Amino acid	ee_L (%) $\pm 3\sigma$	R_s
Proteinogenic amino acids		
Alanine	3.16 ± 0.80	4.00
Aspartic acid	4.31 ± 0.59	2.25
Valine	4.88 ± 0.73	3.20
Glutamic acid	3.79 ± 1.07	3.67
Isoleucine	9.49 ± 1.16	9.00
Leucine	26.33 ± 0.76	4.75
non-proteinogenic amino acids		
2-Aminobutyric acid	2.04 ± 0.86	3.80
3-Aminobutyric acid	5.95 ± 0.62	1.43
Norvaline	0.55 ± 0.21	4.50
Pyroglutamic acid	3.85 ± 0.78	3.20
non-proteinogenic α-dialkyl amino acids		
Isovaline	4.61 ± 0.83	1.67
Methylpyroglutamic acid	0.61 ± 0.03	3.60
Methylleucine	6.16 ± 0.26	2.00



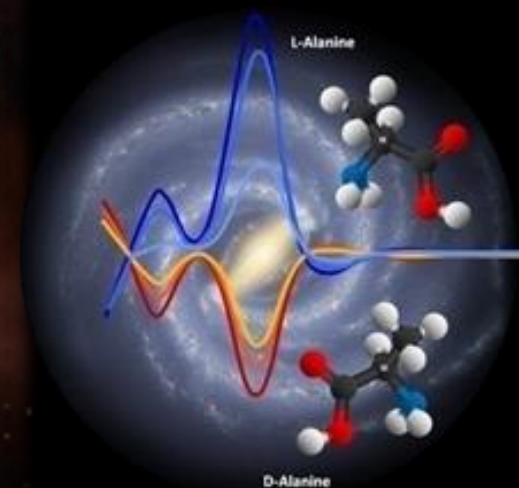
Resolution of amino acid enantiomers in a sample of Murchison using **GCxGC-TOFMS**.

Angew. Chem. Int. Ed. **54** (2015), 1402–1412.

ChemPlusChem **79**, 781–785 (2014).

J. Chromatogr. A **1433**, 131–136 (2016).

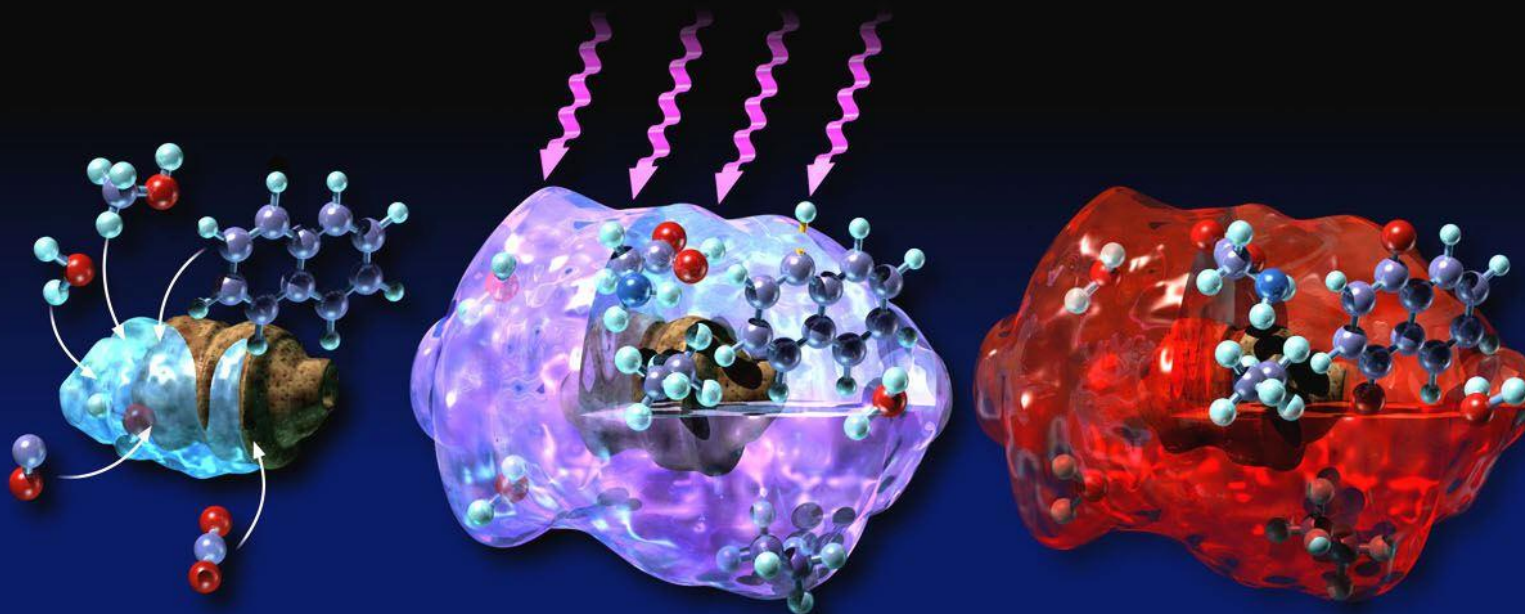
SELECTION OF HANDENESS AT THE DAWN OF MOLECULAR EVOLUTION



Chem. Soc. Rev. 2012

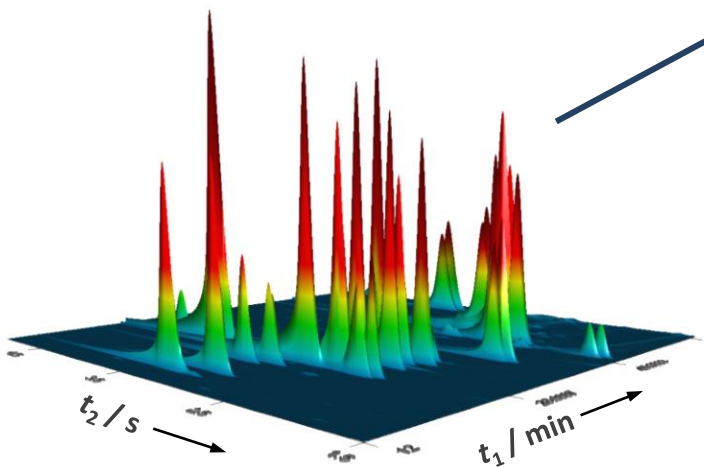
Bailey et al. *Science* **281** (1998)
Kwon et al.: *ApJL* **765** (2013)
Kwon et al.: *Astron J.* **156** (2018)

INTERSTELLAR SYNTHESIS OF CHIRAL MOLECULES

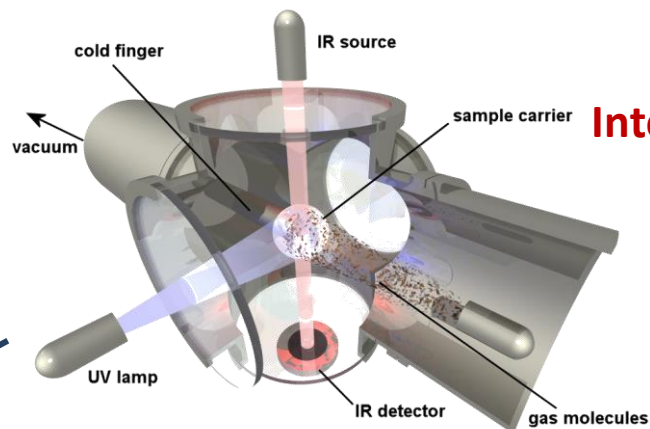


I – CHIRAL BIOMOLECULES IN EXTRATERRESTRIAL SAMPLES

Enantioselective multidimensional gas chromatography



Angew. Chem. 2012



Interstellar ice simulations

PNAS 2015
Science 2016



$ee \neq 0$ Meteorites

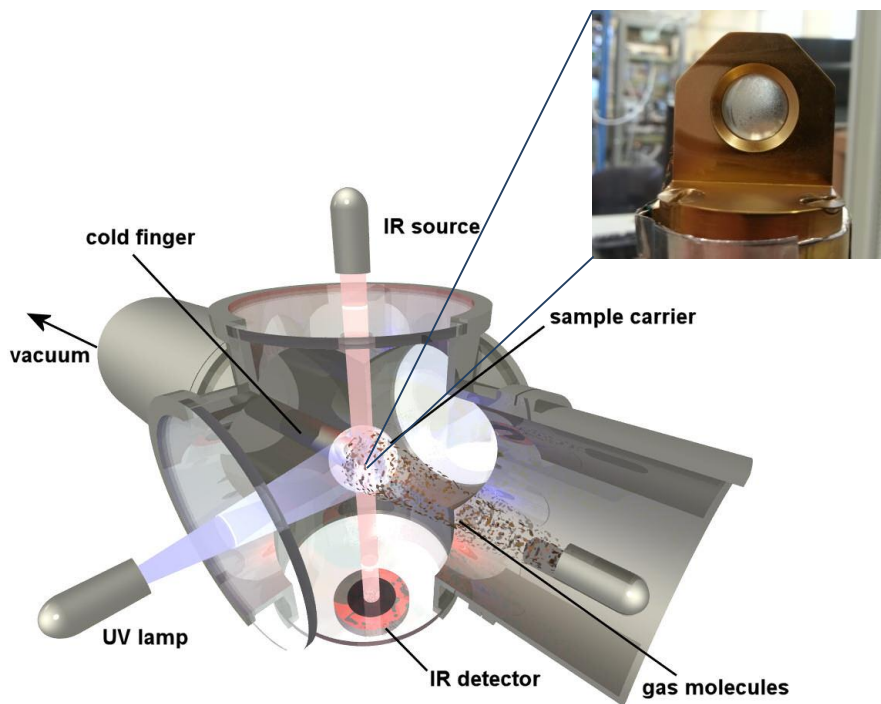
Angew. Chem. 2015



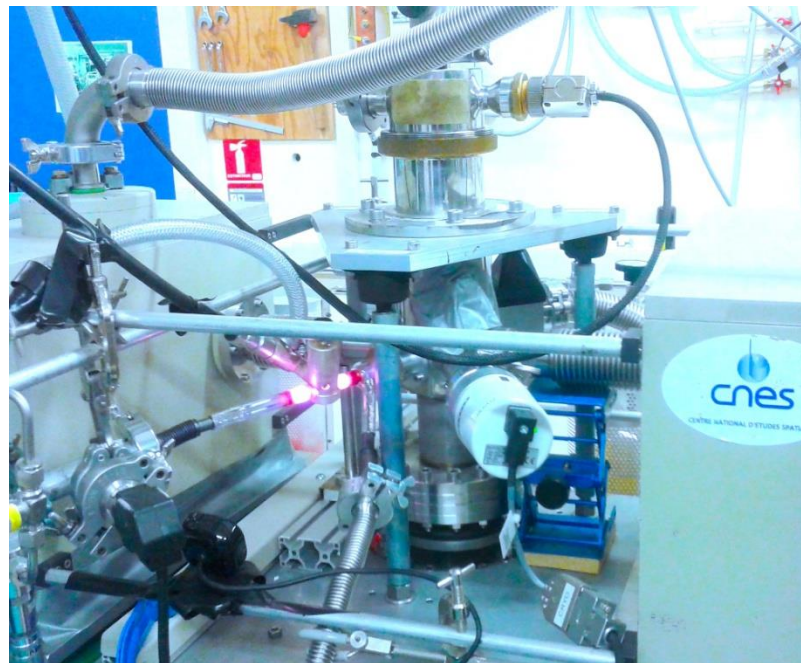
Asteroid sample return missions

Hayabusa2 (JAXA), OsirisRex (NASA)

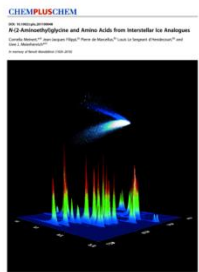
PRIMITIVE CHEMICAL SYNTHESIS: WHAT IS POSSIBLE UNDER SIMULATED INTERSTELLAR CONDITIONS?



Principle of a simulation chamber for interstellar photochemistry: the ice sample composed of H_2O , NH_3 , and $^{13}\text{CH}_3\text{OH}$ is deposited in the center on a MgF_2 -window at a temperature of 80 K and irradiated by vacuum UV light.

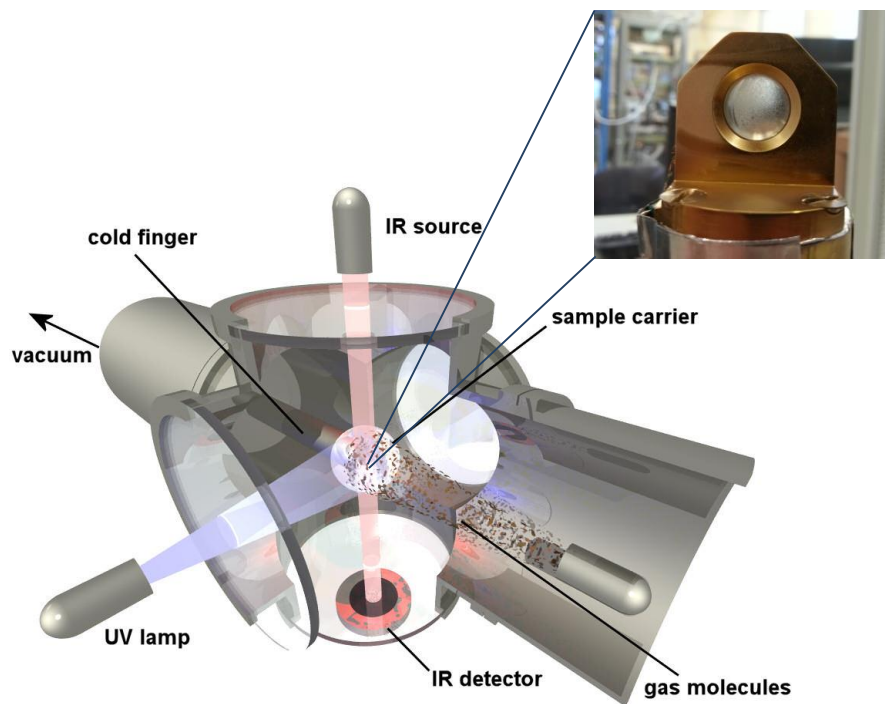


Space simulation chamber @Institut d'Astrophysique Spatiale (IAS), CNRS Université Paris-Sud, Orsay.

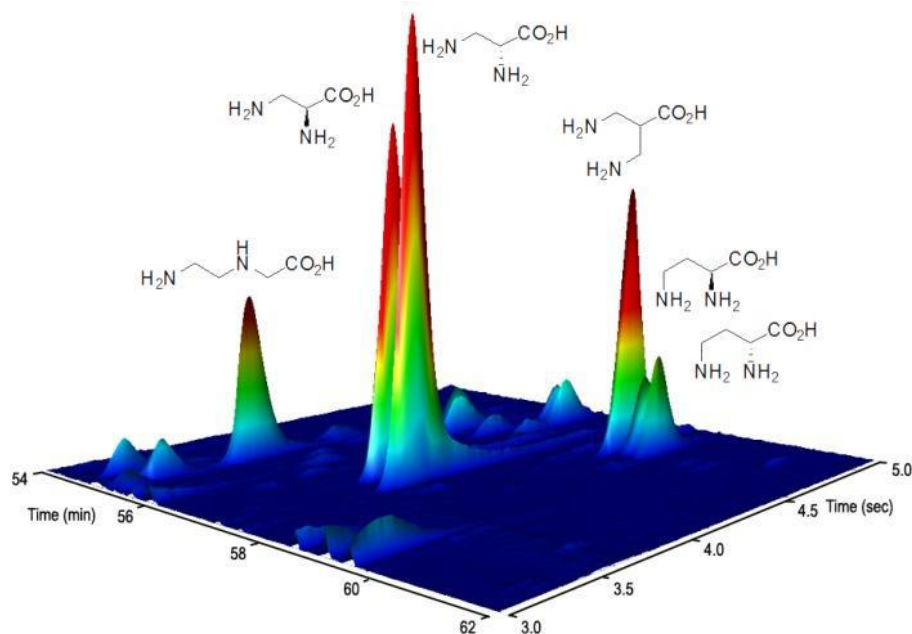


ChemPlusChem **77** (2012), 186–191;
Nature **416** (2002), 403–406.

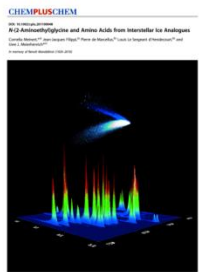
AMINO & DIAMINO ACIDS DETECTED IN COMETARY ANALOGUES



Principle of a simulation chamber for interstellar photochemistry: the ice sample composed of H_2O , NH_3 , and $^{13}\text{CH}_3\text{OH}$ is deposited in the center on a MgF_2 -window at a temperature of 80 K and irradiated by vacuum UV light.

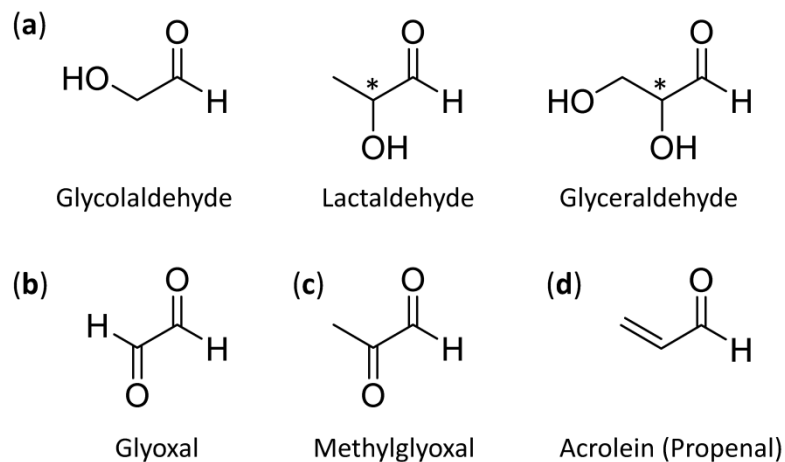


Close-up view of the **2D enantioselective gas chromatogram** depicting **ECHFBE derivatives of ^{13}C -labelled diamino acids**.

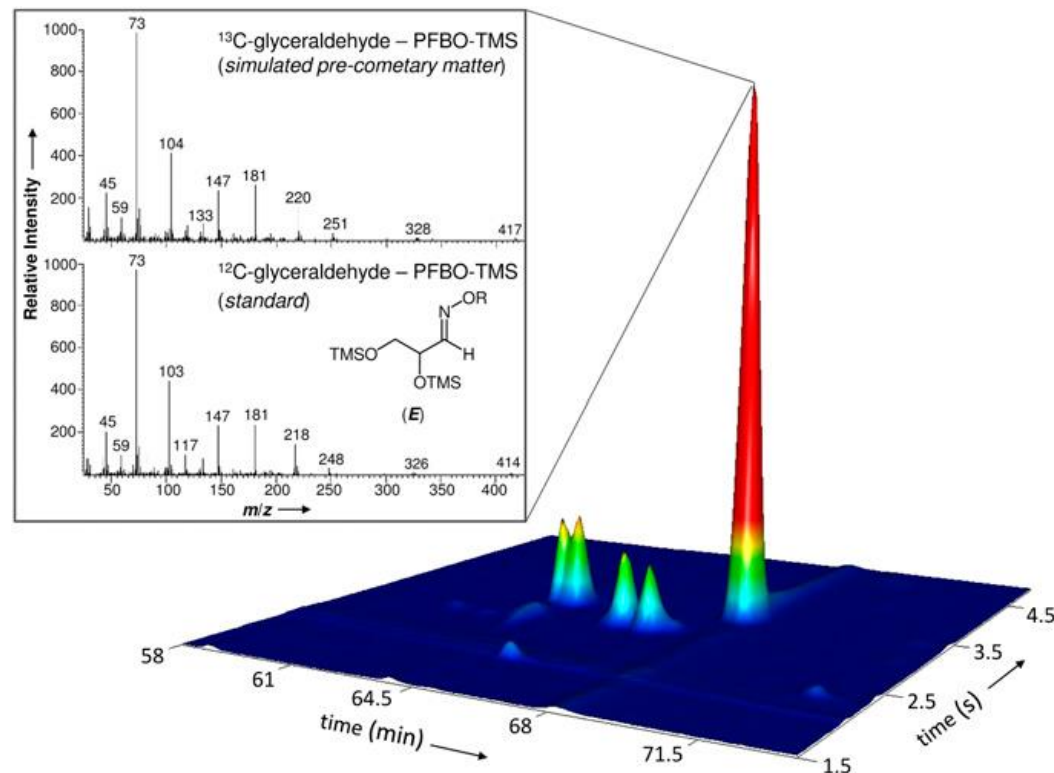


ChemPlusChem **77** (2012), 186–191;
Nature **416** (2002), 403–406.

ALDEHYDES DETECTED IN COMETARY ANALOGUES



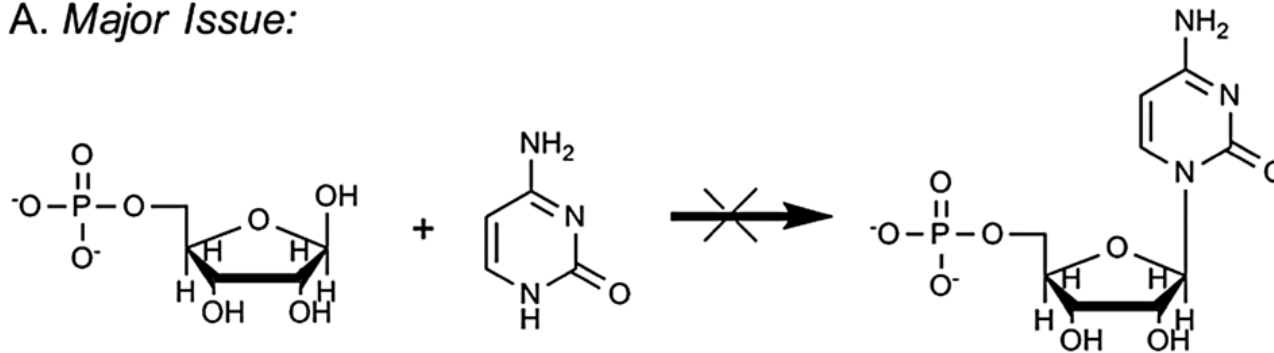
Selected aldehydes identified at room temperature in simulated pre-cometary organic residues.



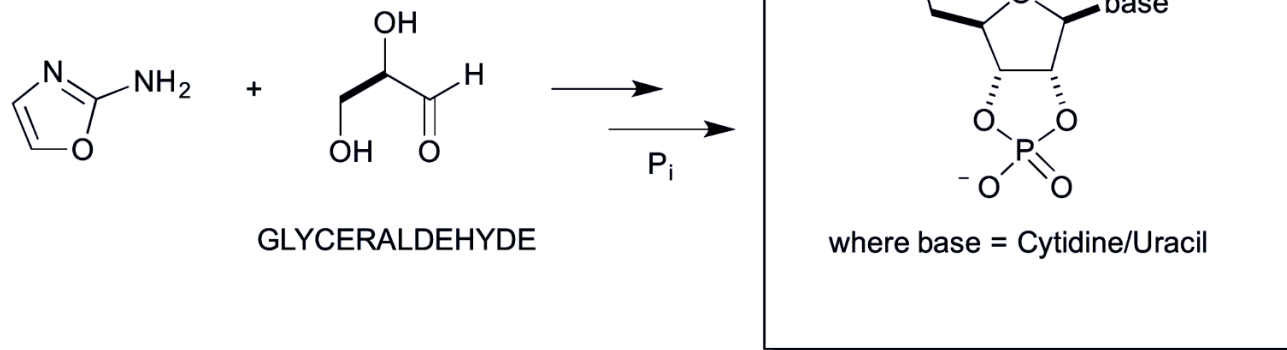
Close-up view of the 2D enantioselective gas chromatogram depicting ECHFBE derivatives of ¹³C-labelled diamino acids.

PNAS **112** (2015), 965–970.

A. Major Issue:



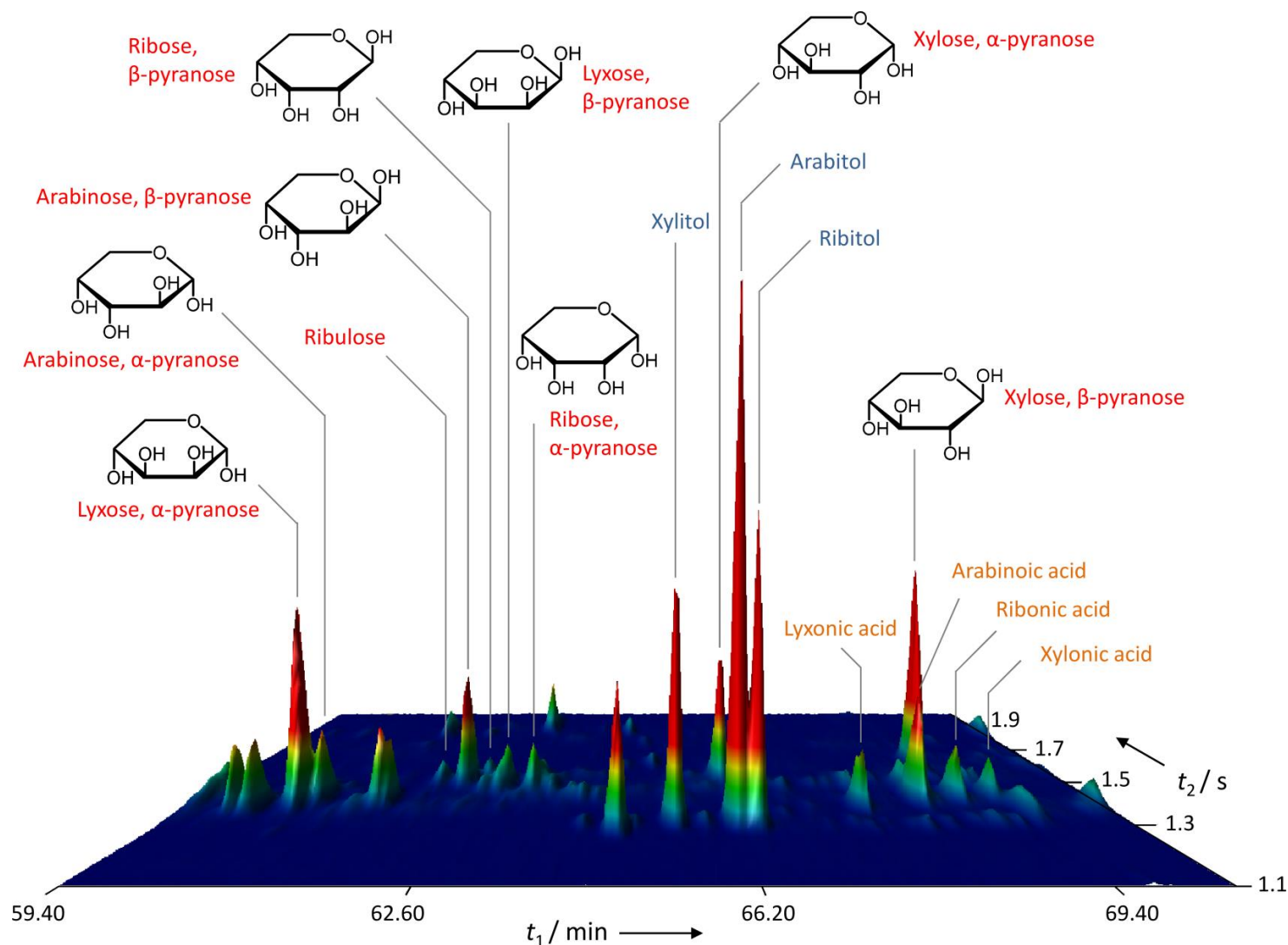
B. Alternative Synthetic Solution:



(A) The RNA World is thermodynamically unfavorable if originating from ribose sugar plus nucleobase; (B) an alternative route to an RNA World.

Powner, Gerland, Sutherland. *Nature* **459** (2009), 239–242

RIBOSE DETECTED IN COMETARY ANALOGUES



RIBOSE - A key sugar found in RNA has been created in the laboratory under conditions similar to those around comets.

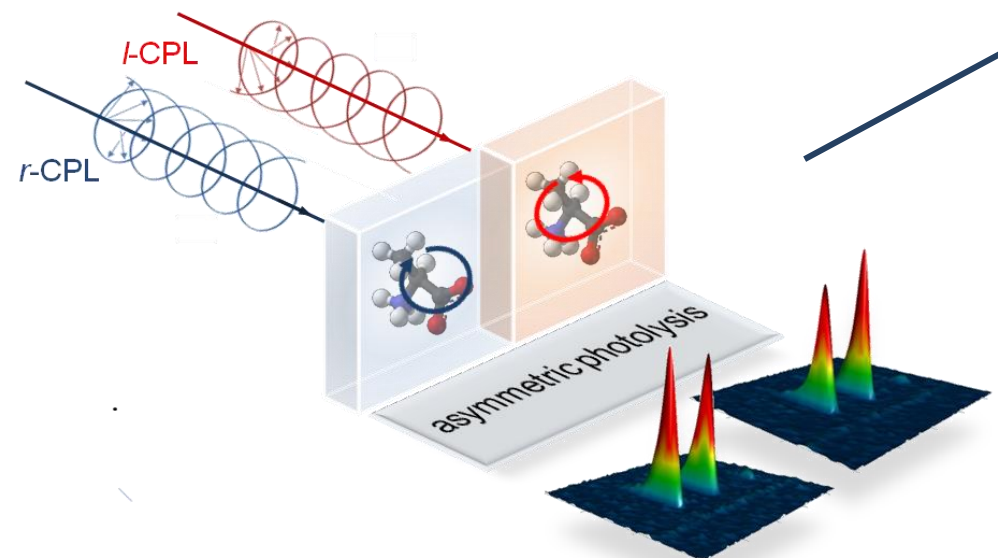
Science **352** (2016), 208–212.

RIBOSE & ITS SISTER ALDOPENTOSE DETECTED IN COMETARY ANALOGUES

	Aldoses				Ketoses	
C-2	$\begin{array}{c} \text{R} \\ \\ \text{CH}_2\text{OH} \end{array}$ R = CH₂OH, Ethylene glycol (550 ppm) = CHO, Glycolaldehyde (2390 ppm) = COOH, Glycolic acid (6330 ppm)					
C-3	$\begin{array}{c} \text{R} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ R = CH₂OH, Glycerol (2860 ppm) = CHO, Glyceraldehyde (302 ppm) = COOH, Glyceric acid (2440 ppm)				$\begin{array}{c} \text{CH}_2\text{OH} \\ \\ \text{C}=\text{O} \\ \\ \text{CH}_2\text{OH} \end{array}$ Dihydroxyacetone (540 ppm)	
C-4	$\begin{array}{c} \text{R} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ R = CH₂OH, Erythritol (5070 ppm) = CHO, Erythrose (< q.l.) = COOH, Erythronic acid (960 ppm)		$\begin{array}{c} \text{R} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ R = CH₂OH, Threitol (7200 ppm) = CHO, Threose (< d.l.) = COOH, Threonic acid (840 ppm)		$\begin{array}{c} \text{CH}_2\text{OH} \\ \\ \text{C}=\text{O} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ Erythrulose (37 ppm)	
C-5	$\begin{array}{c} \text{R} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ R = CH₂OH, Ribitol (560 ppm) = CHO, Ribose (260 ppm) = COOH, Ribonic acid (82 ppm)	$\begin{array}{c} \text{R} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ Arabitol (1150 ppm) Arabinose (200 ppm) Arabinoic acid (165 ppm)	$\begin{array}{c} \text{R} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ Xylitol (630 ppm) Xylose (240 ppm) Xylonic acid (67 ppm)	$\begin{array}{c} \text{R} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ Arabitol (1150 ppm) Lyxose (145 ppm) Lyxonic acid (140 ppm)	$\begin{array}{c} \text{CH}_2\text{OH} \\ \\ \text{C}=\text{O} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ Ribulose (2010 ppm)	$\begin{array}{c} \text{CH}_2\text{OH} \\ \\ \text{C}=\text{O} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OH} \end{array}$ Xylulose (470 ppm)



Asymmetric photolysis / photosynthesis



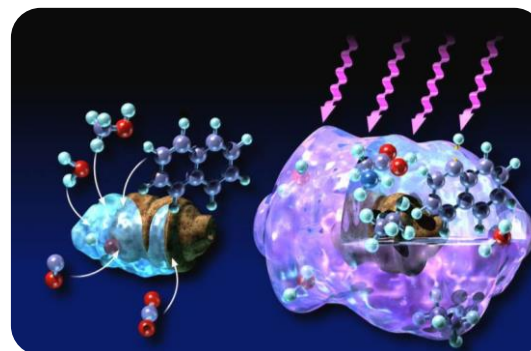
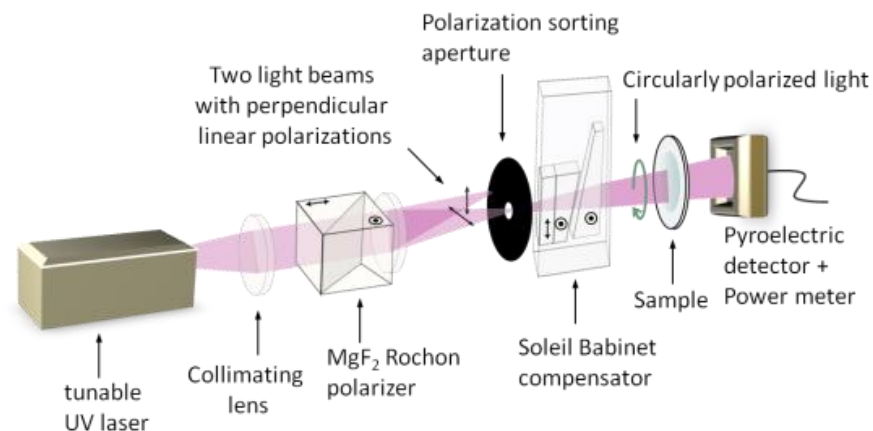
Enantiospecific photochemistry depends on:

$$\Delta\varepsilon = \varepsilon_R - \varepsilon_L$$

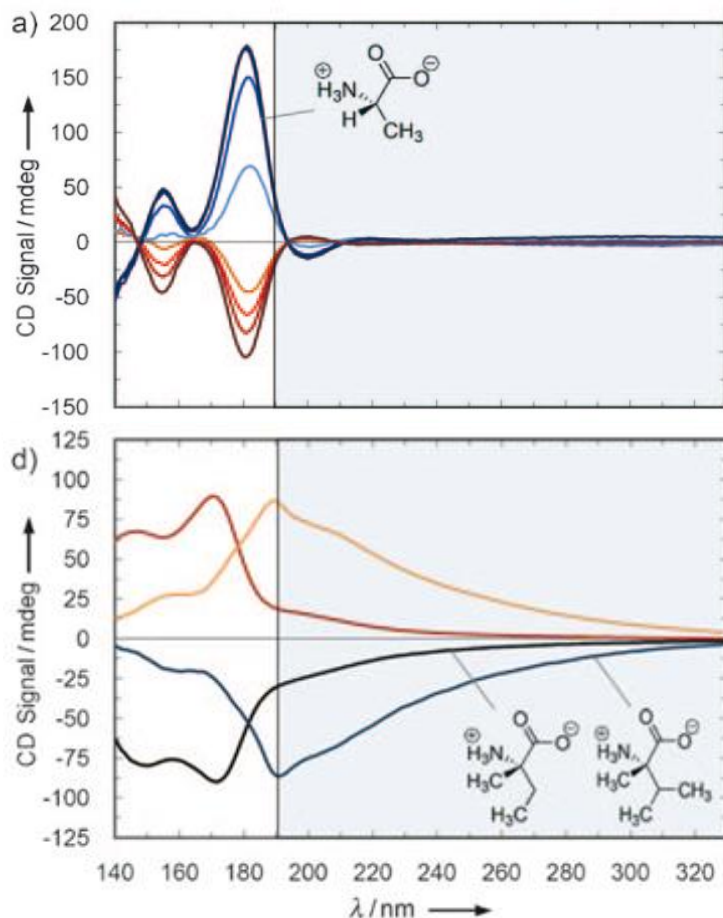
Absolute asymmetric photosynthesis of chiral genome & proteome precursors

in-house energy-tunable laser

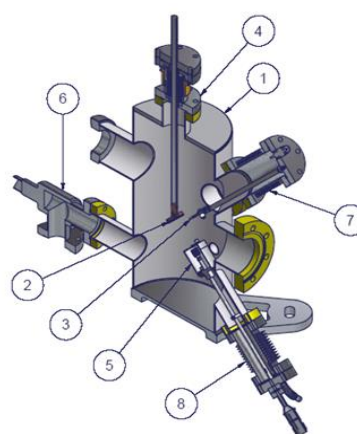
CPL-irradiation experiments....coming soon



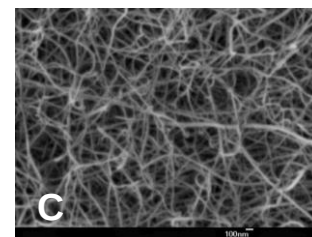
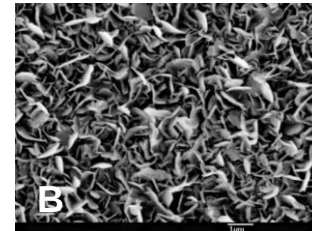
VUV CIRCULAR DICHROISM SPECTRA OF SOLID STATE AMINO ACIDS



CD spectra of isotropic amorphous amino acid films between 140 and 190 nm. a) Blue lines: L-alanine; red lines: D-alanine. b) L-isovaline (black), D-isovaline (red), L-α-methyl valine (blue), D-α-methyl valine (orange).

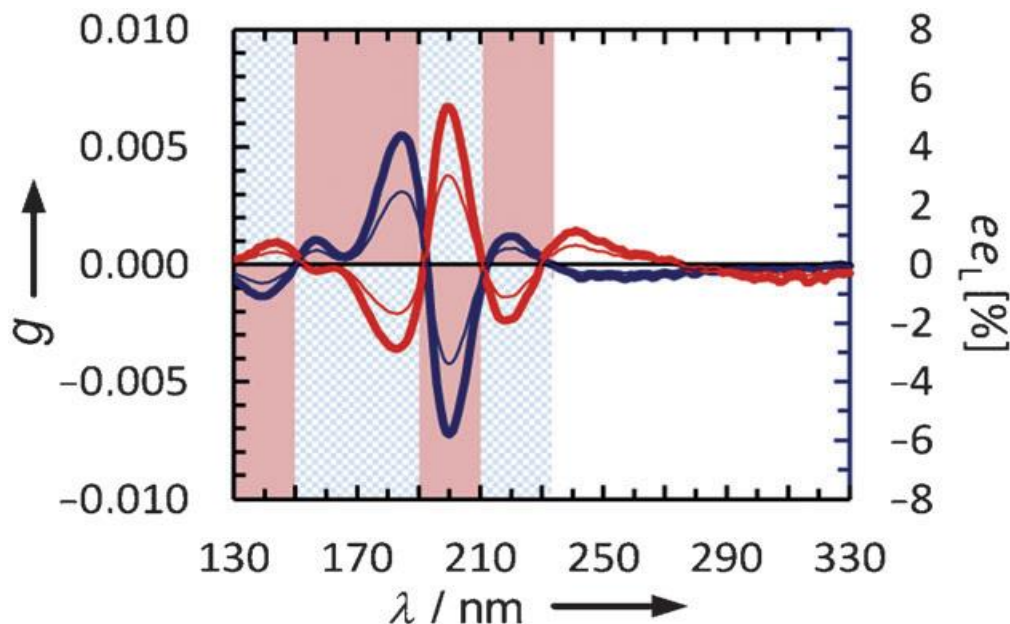


A



Temperature- and pressure-controlled UHV chamber for controlled sublimation of amino acids (A). **Scanning electron microscope (SEM) images** of isotropic amorphous L-valine and (B) D-α-methyl valine (C). The non-crystalline **isotropic amorphous films** show no long-range order.

Angew. Chem. Int. Ed. **49** (2010), 7799–7802.



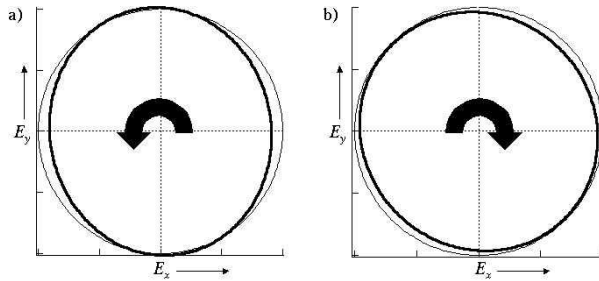
Anisotropy spectra of isotropic amorphous D-alanine (red) and L-alanine (blue) in the vacuum UV and UV spectral region. Thin lines represent the L-enantiomeric excess (ee_L) inducible by either left- or right-cpl into alanine at a given photolysis rate of 99.99 %.

Angew. Chem. Int. Ed. **51** (2012), 4484–4487

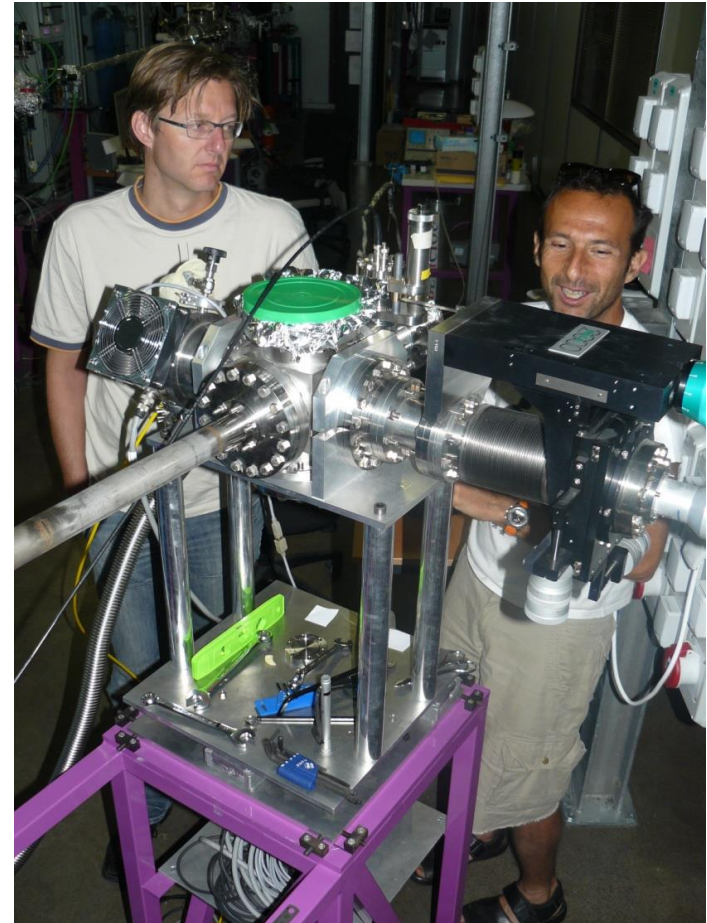
$$g(\lambda) = 4R/D = \Delta\epsilon/\epsilon$$

$$ee \geq (1 - (1 - \xi)^{g/2}) \times 100 \%$$

- a) prediction of the sign of the induced ee ,
- b) determination of kinetics and ees of enantioselective photolysis,
- c) the selection of the CPL wavelength best suited for inducing ee .

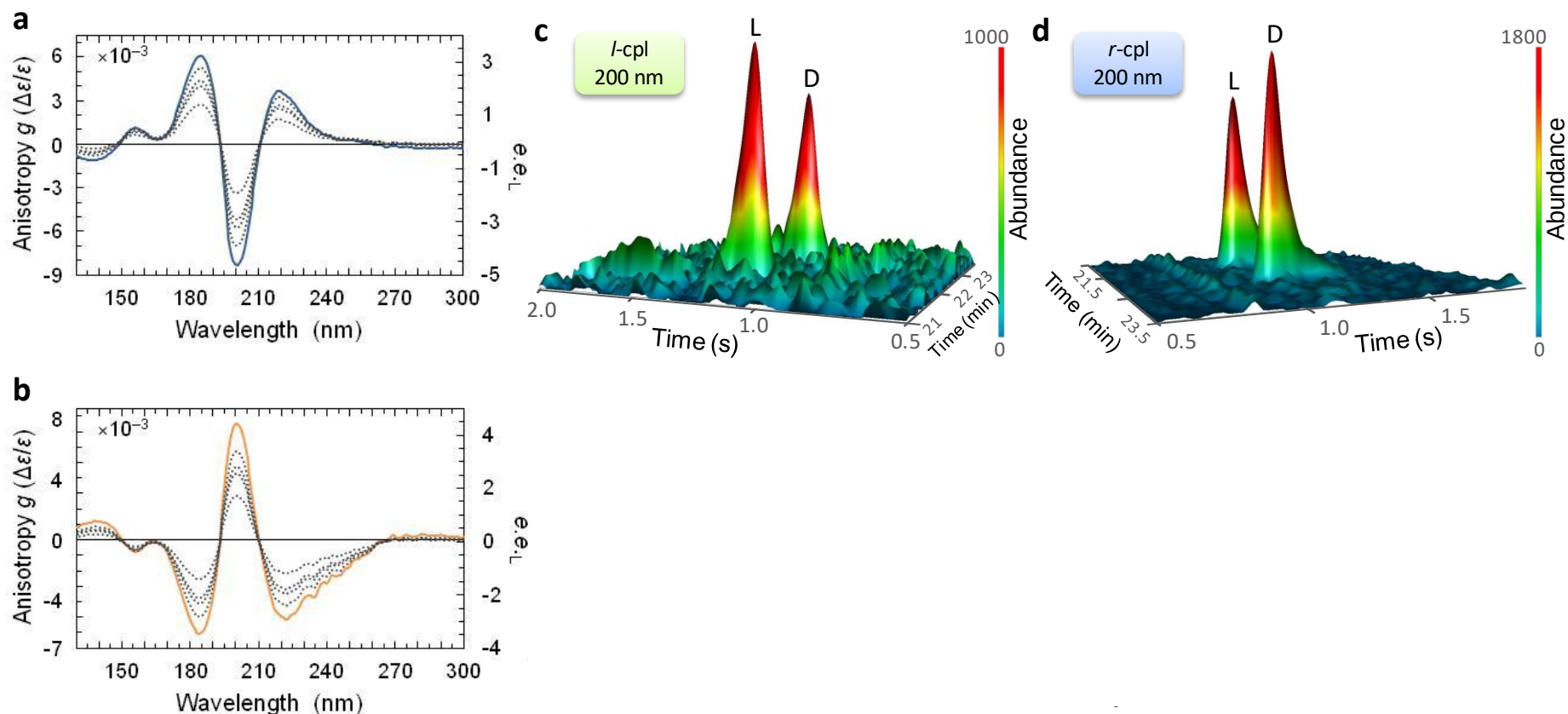


Measured polarization ellipses (thick lines) at the sample location as produced by the OPHELIE2, HU640 type undulator. Absolute circular polarization rates are of 91 % for r-CPSR (b) and 94 % for l-CPSR (a).



Synchrotron SOLEIL is a third generation synchrotron optimized for intense photon fluxes.

PHOTON-ENERGY-CONTROLLED SYMMETRY BREAKING

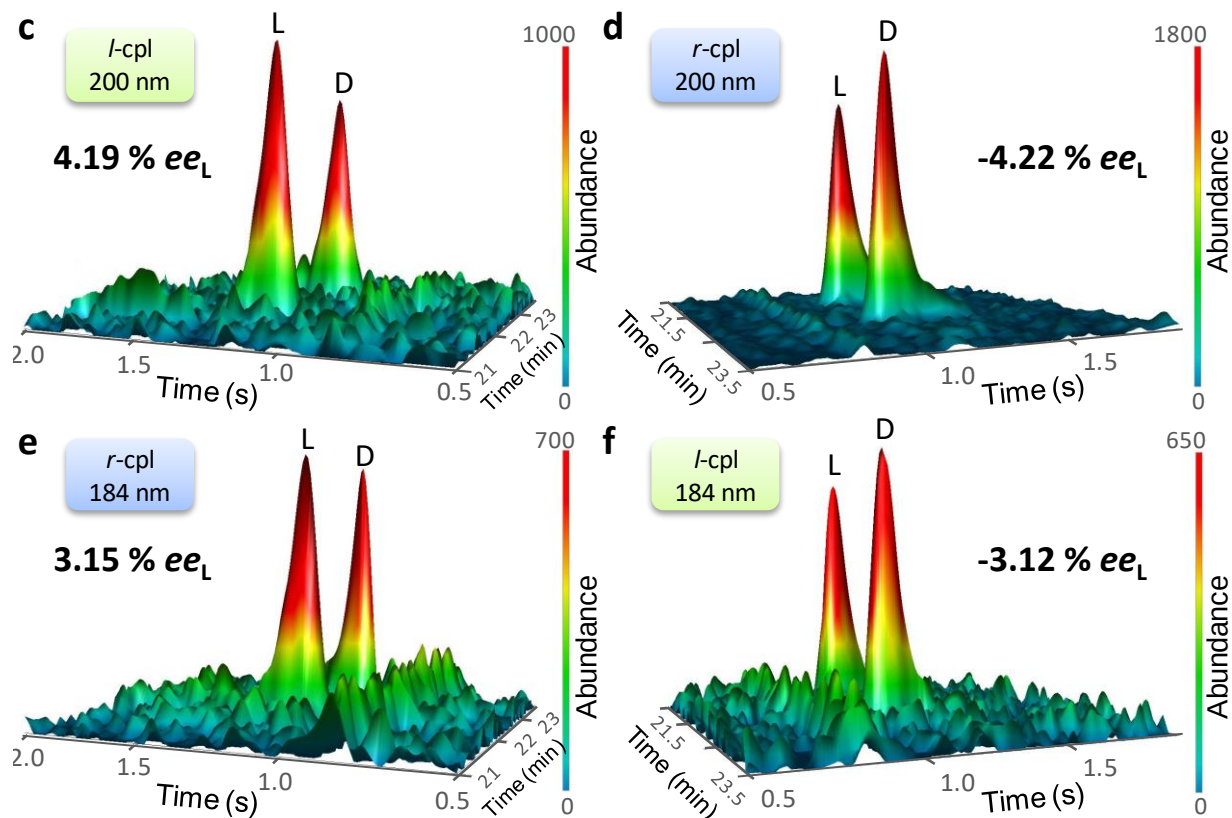


Energy- and polarization-dependent CPL-induced symmetry breaking. (a, b) Anisotropy spectra $g(\lambda)$ and inducible ee_L of L-Ala & D-Ala. (c–d) Enantioselective GCxGC of ^{13}C -Ala after irradiation with different CPL-polarization and at different photon energies for 5h.

Angew. Chem. Int. Ed. **53** (2014), 210–214

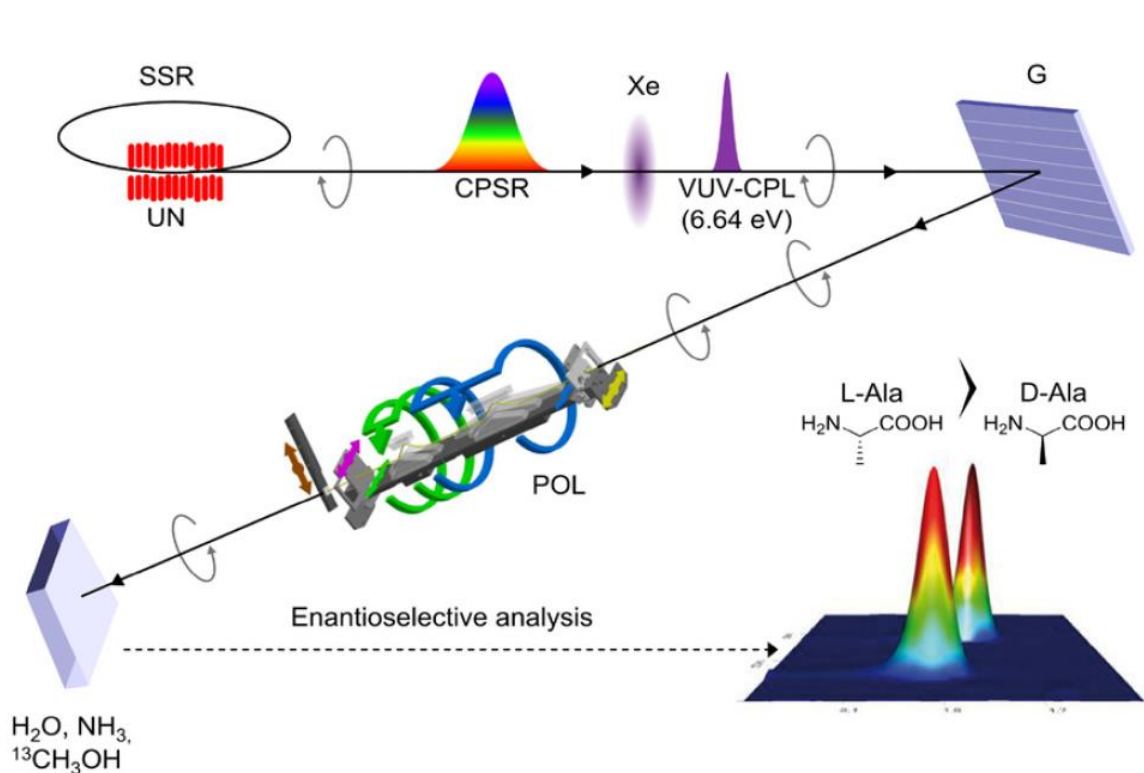
➤ sign of induced ee depends upon helicity of CPL

➤ sign of induced ee depends upon the energy of l -CPL and r -CPL



Energy- and polarization-dependent CPL-induced symmetry breaking. (a, b) Anisotropy spectra $g(\lambda)$ and inducible ee_L of L-Ala & D-Ala. (c–d) Enantioselective GCxGC of ^{13}C -Ala after irradiation with different CPL-polarization and at different photon energies for 5h.

ABSOLUTE ASYMMETRIC AMINO ACID SYNTHESIS

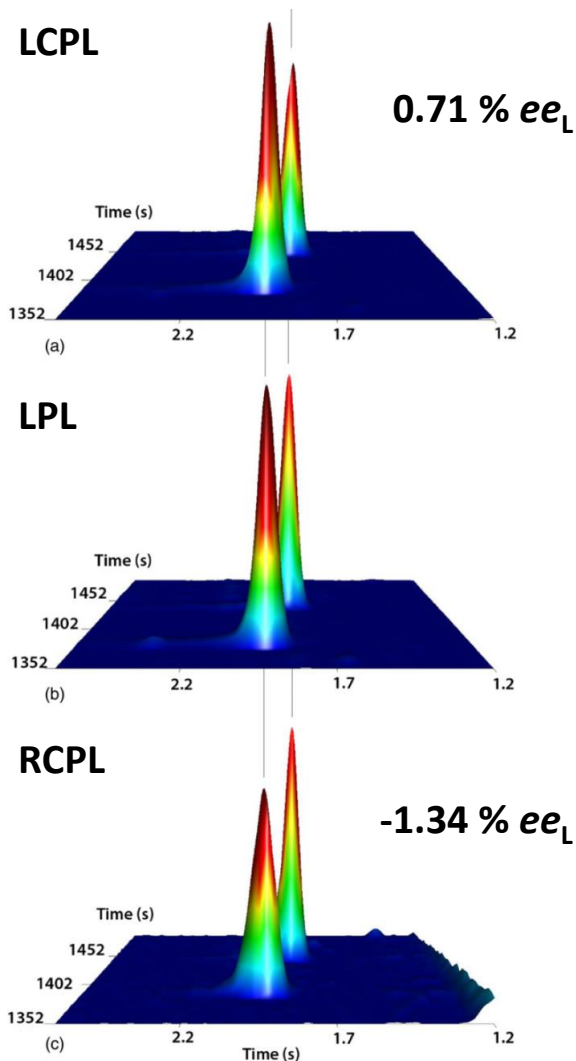


Set-up of asymmetric VUV photolysis at the synchrotron radiation facility SOLEIL (left). Multidimensional gas chromatograms of ¹³C-alanine enantiomers for the three polarization regimes (right).

Astroph. J. Letters **727** (2011), L27

Follow-up study: asymmetric synthesis of five amino acids

Astroph. J. Letters **727** (2014) L27.



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Dr. L. NAHON

Beamline DESIRS
Synchrotron SOLEIL
Gif-sur-Yvette, Fr



***CHIRALITY as a trigger
of selection,
interaction &
feedback in the
evolution of the first
biopolymers***



April 1st, 2019

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